SUBSTRATES TO STUDY THE MECHANISM OF VITAMIN D HYDROXYLATION: SYNTHESIS OF $[24R-^{2}H]-25-HYDROXYPROVITAMIN D_{2}$

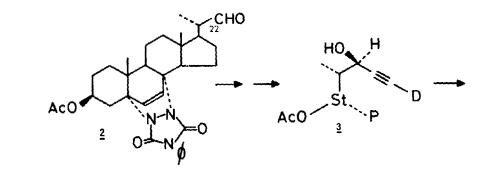
by James D. Meadows and Dudley H. Williams* University Chemical Laboratory, Lensfield Road., Cambridge CB2 1EW, U.K.

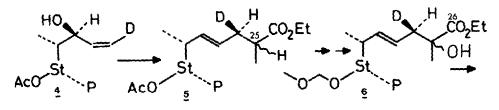
Summary: $[24R^{-2}H]$ -25-Hydroxyprovitamin D₃ has been synthesised, in order that the corresponding vitamin may be used to study the stereochemistry of hydroxylation in production of the metabolite 24R,25-dihydroxycholecalciferol. The stereospecific introduction of deuterium at C-24 is effected by means of a Claisen rearrangement.

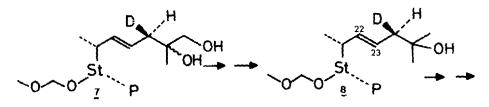
24R,25-Dihydroxycholecalciferol is an important metabolite of vitamin $D_3^{1,2}$ which may be concerned with calcium action in bone.^{3,4} The site of 24hydroxylation is kidney,⁵ but whether hydroxylation occurs with retention or inversion of configuration is unknown. The expectation is that hydroxylation would occur with retention of configuration (insertion of oxygen into the C-H bond).⁶ We report here the synthesis of $[24R-^2H]-25$ -hydroxyprovitamin D_3 (<u>1</u>), suitable for conversion into the corresponding vitamin, and hence for testing the mechanism of hydroxylation.

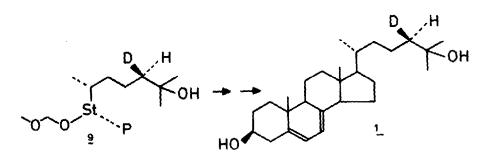
The C-22 aldehyde $\underline{2}$, ⁷ in which the ring B diene has been protected by reaction with phenyltriazolinedione, was prepared from ergosterol acetate. The aldehyde underwent selective attack when treated with 2 equivalents of lithium trimethylsilyacetylide in tetrahydrofuran (THF) at -78° . The reaction was quenched (aqueous NH₄Cl) after 3h, and the resulting mixture of diastereoisomers at C-22 (3:1 ratio) separated by short column chromatography.⁸ The major diastereoisomer (mp 200-201, 50% yield) was shown to possess the 22R stereochemistry by X-ray crystallography of its p-bromobenzoate (mp 198-199).⁹ The synthesis was continued with this diastereoisomer, and the trimethylsilyl group removed, by treatment with 10 equivalents of dry benzyltrimethylammonium

4373









St = steroid nucleus P = ring-B protecting group

fluoride in THF containing 100 molar equivalents of 99.8% D_2O , to give 3. This acetylene was converted to the Z-allylic alcohol 4 by catalytic hydrogenation over Lindlar's catalyst in THF. Upon reaction of 4 with triethylorthopropionate, and propionic acid as catalyst, in refluxing dry benzene for 16h, the Claisen rearrangement product 5 was obtained in 80% yield as a mixture of diastereoisomers at C-25.¹⁰ In order to obtain the quoted yield, it was necessary to titrate the crude product with a small amount of phenyltriazolinedione, to replace a portion of the ring B protecting group which is lost .

At a later stage in the synthesis, a base-stable protecting group is required for the 3 β -OH group.¹¹ Reaction of <u>5</u> with potassium carbonate in dry ethanol gave the corresponding 3 β -alcohol, which afforded the 3 β -methoxymethyl ether upon treatment with chloromethylmethyl ether and N,N-diisopropylethylamine in CH₂Cl₂. This product was converted¹² to the α -hydroxy ester <u>6</u> in 75% yield upon reaction of the C-25 anion (generated by use of 1.2 equivalents of lithium isopropylcyclohexylamide at -78° in THF) with oxygen at -78°, and reduction of the resulting peroxide with triethyl phosphite. The C-26 ester group was then removed by (i) reduction of <u>6</u> to the C-26 alcohol with LiBH₄ in THF to give <u>7</u> (95% yield) and (ii) reduction of the C-26 tosylate (tosyl chloride/pyridine) with LiBH₄ in refluxing THF to give <u>8</u> (70% yield).

Reaction of <u>8</u> with BH_3 (1 equivalent) in THF for 1 h at room temperature gave a C(22)-C(23) adduct which, unfortunately, was not cleaved by acetic acid. The adduct was therefore worked-up oxidatively (H_2O_2 in aqueous sodium carbonate) to give a 23,25-diol, which was converted to the 23-tosylate (tosyl chloride/ pyridine), and this in turn was reduced with LiBH₄ in THF for 12h to give <u>9</u> (20% overall yield from <u>8</u>). Removal of the protecting groups (TsOH in MeOH under reflux for 2h, followed by reaction with LiAlH₄ in THF at 50^o for 12h) gave $[24R-^{2}H]$ -25-hydroxyprovitamin D_3 (<u>1</u>) in 70% overall yield from <u>9</u>.

The problems associated with removal of the C(22)-C(23) double bond detract from this route to introduce chirality at C-24 in a saturated sidechain. However, it is noteworthy that an analogue of $\underline{3}$, in which CH₃ replaced deuterium, would upon LiAlH₄ reduction,¹³ Claisen rearrangement, C-25 hydroxylation, and reduction furnish 25,26-dihydroxyvitamin D₂ in a relatively efficient synthesis. We are currently exploring this route.

Acknowledgements

We thank SRC, the Smith Kline and French Foundation, and Roche Products Ltd., U.K., for support of this work.

References

- 1 M.F. Holick, H.F. Schnoes, H.F. DeLuca, R.W. Gray, I.T. Boyle, and T. Suda, Biochemistry, 11, 4251 (1972).
- 2 Y. Tanaka, H.F. DeLuca, N. Ikekawa, M. Monsaki, and N. Koizumi, <u>Arch</u>. <u>Biochem. Biophys.</u>, <u>170</u>, 620 (1975).
- 3 A. Hay, <u>Nature</u>, <u>278</u>, 509 (1979).
- 4 D. Goodwin, D. Noff, and S. Edelstein, <u>Nature</u>, <u>276</u>, 517 (1978).
- 5 I.T. Boyle, J.L. Omdall, R.W. Gray, and H.F. DeLuca, <u>J. Biol. Chem.</u>, 248, 4174 (1973).
- 6 K.R. Hansen and I.A. Rose, <u>Acc. Chem. Res.</u>, <u>8</u>, 1 (1975).
- 7 D.H.R. Barton, T. Shiori, and D.A. Widdowson, <u>J. Chem. Soc.</u> (C), 1968 (1971).
- 8 All isolated substanced described in this paper were characterised by mass spectrometric and ¹H NMR data which were in full accord with the assigned structures.
- 9 F.H. Allen, W.B.T. Cruse, and O. Kennard, Acta Cryst. (B), in press.
- 10 See, for example, W. Sucrow, M. Slopianka, and P. Caldeira, <u>Chem.</u> <u>Ber.</u>, <u>108</u>, 1101 (1975).
- 11 Such a protecting group could not be introduced at the outset of the synthesis, since separation of a pure C-22 diastereoisomer corresponding to 3 (but, for example, with a methoxymethyl protecting group at 3β -OH)then proved impossible.
- 12 H.H. Wasserman and B.H. Lipshutz, Tetrahedron Letters, 1731 (1975).
- 13 R.A. Raphael, "Acetylenic Compounds in Organic Synthesis", Butterworth and Co., Ltd., London 1955, p.29.

(Received in UK 8 August 1980)